

Rietveld refinement of langbeinite-type $K_2YHf(PO_4)_3$

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Key indicators: powder X-ray study; $T = 293$ K; mean $\sigma(P-O) = 0.017$ Å; disorder in main residue; R factor = 0.054; wR factor = 0.071; data-to-parameter ratio = 11.0.

Potassium yttrium hafnium tris(orthophosphate) belongs to the langbeinite-family and is built up from $[MO_6]$ octahedra [in which the positions of the two independent M sites are mutually occupied by Y and Hf in a 0.605 (10):0.395 (10) ratio] and $[PO_4]$ tetrahedra connected *via* vertices into a three-dimensional framework. This framework is penetrated by large closed cavities in which the two independent K atoms are located; one of the K atoms is nine-coordinated and the other is 12-coordinated by surrounding O atoms. The K, Y and Hf atoms lie on threefold rotation axes, whereas the P and O atoms are located in general positions.

Related literature

For the structure of the mineral langbeinite, see: $K_2Mg_2(SO_4)_3$ (Zemann & Zemann, 1957). For powder diffraction investigations and Rietveld refinements of phosphate-based langbeinites, see: $K_2MZr(PO_4)_3$, $M = Y, Gd$ (Wulff *et al.*, 1992); $K_2FeZr(PO_4)_3$ (Orlova *et al.*, 2003); $K_2LnZr(PO_4)_3$, $Ln = Ce-Lu$ (Trubach *et al.*, 2004). Hafnium-containing phosphate langbeinites are reported for $K_2BiHf(PO_4)_3$ (Losilla *et al.*, 1998) and $K_{1.93}Mn_{0.53}Hf_{1.47}(PO_4)_3$ (Ogorodnyk *et al.*, 2007a). For the synthesis of zirconium- or hafnium-containing langbeinite-related phosphates from fluoride precursors using flux techniques, see: Ogorodnyk *et al.* (2007a,b). Parameters needed to calculate bond-valence sums were taken from Brown & Altermatt (1985) and Brese & O'Keeffe (1991), respectively. For ionic radii, see: Shannon (1976). For crystallographic background, see: Boultif & Louër (2004).

Experimental

Crystal data

$K_2YHf(PO_4)_3$
 $M_r = 630.51$
Cubic, $P2_13$
 $a = 10.30748$ (9) Å

$V = 1095.11$ (2) Å³
 $Z = 4$
Cu $K\alpha$ radiation
 $T = 293$ K

Specimen shape: flat sheet
 $15 \times 15 \times 1$ mm
Specimen prepared at 101.3 kPa

Specimen prepared at 293 K
Particle morphology: isometric, colourless

Data collection

Shimadzu XRD-6000 diffractometer
Specimen mounting: glass container
Specimen mounted in reflection mode

Scan method: step
 $2\theta_{\min} = 5.0$, $2\theta_{\max} = 105.0^\circ$
Increment in $2\theta = 0.02^\circ$

Refinement

$R_p = 5.375$
 $R_{wp} = 7.075$
 $R_{exp} = 2.809$
 $R_B = 4.248$
 $S = 2.51$
Wavelength of incident radiation: 1.540530 Å

Profile function: Thompson-Cox-Hastings pseudo-Voigt with axial divergence asymmetry (Thompson *et al.*, 1987)
528 reflections
48 parameters

Table 1

Selected geometric parameters (Å, °). $M = Hf, Y$.

K1—O1 ⁱ	2.981 (16)	M1—O2 ^{iv}	2.085 (15)
K1—O2 ⁱⁱ	3.345 (14)	M2—O3 ⁱ	2.211 (14)
K1—O4 ⁱⁱ	3.413 (16)	M2—O4	2.113 (17)
K2—O3 ⁱⁱ	2.907 (14)	P1—O1	1.518 (16)
K2—O2 ⁱⁱⁱ	2.912 (14)	P1—O2	1.621 (17)
K2—O4 ⁱⁱ	3.207 (17)	P1—O3	1.470 (16)
K2—O4 ⁱⁱⁱ	3.336 (17)	P1—O4	1.497 (19)
M1—O1	2.148 (14)		
O1—P1—O2	100.5 (9)	O2—P1—O3	121.4 (8)
O1—P1—O3	113.0 (9)	O2—P1—O4	107.7 (9)
O1—P1—O4	106.9 (9)	O3—P1—O4	106.5 (9)

Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + \frac{3}{2}, -y + 1, z + \frac{1}{2}$; (iii) $-z + 1, x + \frac{1}{2}, -y + \frac{3}{2}$; (iv) $x - \frac{1}{2}, -y + \frac{1}{2}, -z$.

Data collection: *PCXRD* (Shimadzu, 2006); cell refinement: *FULLPROF* (Rodríguez-Carvajal, 2006); data reduction: *FULLPROF*; method used to solve structure: coordinates taken from an isotopic structure; program(s) used to refine structure: *FULLPROF*; molecular graphics: *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *PLATON* (Spek, 2009) and *enCIFer* (Allen *et al.*, 2004).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WM2244).

References

- Allen, F. H., Johnson, O., Shields, G. P., Smith, B. R. & Towler, M. (2004). *J. Appl. Cryst.* **37**, 335–338.
Boultif, A. & Louër, D. (2004). *J. Appl. Cryst.* **37**, 724–731.
Brandenburg, K. (1999). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
Brese, N. E. & O'Keeffe, M. (1991). *Acta Cryst.* **B47**, 192–197.
Brown, I. D. & Altermatt, D. (1985). *Acta Cryst.* **B41**, 244–247.
Losilla, E. R., Bruque, S., Aranda, M. A. G., Moreno-Real, L., Morin, E. & Quarton, M. (1998). *Solid State Ionics*, **112**, 53–62.
Ogorodnyk, I. V., Baumer, V. N., Zatovsky, I. V., Slobodyanik, N. S., Shishkin, O. V. & Domasevitch, K. V. (2007b). *Acta Cryst.* **B63**, 819–827.
Ogorodnyk, I. V., Zatovsky, I. V., Baumer, V. N., Slobodyanik, N. S., Shishkin, O. V. & Vorona, I. P. (2007a). *J. Solid State Chem.* **180**, 2838–2844.
Orlova, A. I., Trubach, I. G., Kurazhkovskaya, V. S., Pertierra, P., Salvado, M. A., Garcia-Granda, S., Khainakov, S. A. & Garcia, J. R. (2003). *J. Solid State Chem.* **173**, 314–318.
Rodríguez-Carvajal, J. (2006). *FULLPROF*. Laboratoire Le'on Brillouin (CEA-CNRS), France.

- Shannon, R. D. (1976). *Acta Cryst.* **A32**, 751–767.
- Shimadzu (2006). *PCXRD*. Shimadzu Corporation, Kyoto, Japan.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Thompson, P., Cox, D. E. & Hastings, J. B. (1987). *J. Appl. Cryst.* **20**, 79–83.
- Trubach, I. G., Beskrovnyi, A. I., Orlova, A. I., Orlova, V. A. & Kurazhkovskaya, V. S. (2004). *Crystallogr. Rep.* **49**, 614–618.
- Wulff, H., Guth, U. & Loescher, B. (1992). *Powder Diffr.* **7**, 103–106.
- Zemann, A. & Zemmann, J. (1957). *Acta Cryst.* **10**, 409–413.

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Comment

Among a great variety of langbeinite-type based phosphate (mineral langbeinite $K_2Mg_2(SO_4)_3$, Zemann & Zemmann, 1957), only several compounds containing hafnium were reported: $K_2BiHf(PO_4)_3$ (Losilla *et al.*, 1998) and $K_{1.93}Mn_{0.53}Hf_{1.47}(PO_4)_3$ (Ogorodnyk *et al.*, 2007a). At the same time a great number of zirconium-containing phosphates with langbeinite framework were synthesized and structurally characterized: $K_2MZr(PO_4)_3$, $M = Y, Gd$ (Wulff *et al.*, 1992); $K_2FeZr(PO_4)_3$ (Orlova *et al.*, 2003); $K_2LnZr(PO_4)_3$, $Ln = Ce-Lu$ (Trubach *et al.*, 2004b); $Cs_{1+x}Ln_xZr_{2-x}(PO_4)_3$, $Ln = Sm-Lu$ (Ogorodnyk *et al.*, 2007b). This can be connected with the similarity of chemical behavior of zirconium and hafnium on the one hand and the rareness (or, possibly, the high prices) of hafnium raw materials in comparison with zirconium ones on the other hand.

Due to similar chemical properties of Zr and Hf and the close values of their ionic radii (for coordination number 6 they are 0.72 and 0.71 Å for Zr and Hf, respectively; Shannon, 1976) the cell parameters of $K_2YHf(PO_4)_3$ are slightly smaller than of $K_2YZr(PO_4)_3$ ($a = 10.3346$ (1) Å; Wulff *et al.*, 1992).

K, Y and Hf atoms lie on the 3-fold rotation axes in $4a$ positions (Fig. 1). P and O atoms are located in $12b$ positions. Both Y and Hf atoms occupy two hexacoordinated positions competitively. $M1$ position is preferably occupied by Hf while $M2$ is by Y. The structure contains $[MO_6]$ octahedra and $[PO_4]$ tetrahedra which are connected *via* vertices. Two nearest $[MO_6]$ octahedra are joined to each other by three bridging orthophosphate tetrahedra forming $\{M_2P_3O_{18}\}$ groups. These groups form three-dimensional framework penetrated with large closed cavities. Two independent potassium atoms are located in each cavity. K1 atom is nine-coordinated, while K2 is twelve-coordinated.

Bond valence sums (BVS) were calculated using parameters for Hf, Y, P from Brese & O'Keeffe (1991) and for K from Brown & Altermatt (1985). The calculation were performed for formula sum $K_2YHf(PO_4)_3$ taking into account occupancies of the octahedrally coordinated M positions. The sum of BVS of positively charged atoms is equal to 24.16 while the chemical charge of the remaining O atoms is equal to -24.

Experimental

Well-shaped tetrahedral crystals of $K_2YHf(PO_4)_3$ were grown using a flux technique. A mixture of 4.52 g KPO_3 and 3.4 g $K_4P_2O_7$ (initial K/P molar ratio was set equal to 1.35) was melted in a platinum crucible at 1273 K. The melt was kept at this temperature for 1 h and after that the temperature was decreased to 1173 K. Dispersed in an agate mortar, a mixture of 1.36 g HfF_4 and 0.78 g YF_3 was added to the phosphate flux under stirring. The crystallization of the melts was performed from 1173 to 893 K at a rate of 30 K/h. The synthesized crystalline sample was separated from remaining glass by leaching with hot water. The dimensions of the crystals were found to be in a range 0.01–0.05 mm. The sample was ground in an agate mortar before performing powder XRD data collection. The recorded powder pattern indicated a single phase material.

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The element ratio was determined using ICP-AES analyses (Shimadzu ICPE-9000 spectrometer). The sample for measurements was prepared by dissolution of calculated amount of $\text{K}_2\text{YHf}(\text{PO}_4)_3$ in sulfuric acid (98%) with final dilution by bidistilled water. Element ratio was found to be: 2.02:0.97:0.98:3.04 for K:Y:Hf:P which fits well with the theoretical values.

Refinement

The cubic cell was found by Dicvol 2004 (Boultif & Louër, 2004). The Hf-containing langbeinite-related compound with general composition $\text{K}_{1.93}\text{Mn}_{0.53}\text{Hf}_{1.47}(\text{PO}_4)_3$ (ICSD-418669, Ogorodnyk *et al.*, 2007a) was selected as a starting model for Rietveld refinement. At first profile matching refinement was performed. Then background and scaling factors were added to the refined parameters. The background was approximated using a 6-coefficient polynomial function. Modified pseudo-Voigt function (Thompson *et al.*, 1987) was used for the profile refinement. On the next stage atomic positions were refined. Due to previous investigations of langbeinite-related phosphates and close ionic radii of Y and Hf common positions *M1* and *M2* occupied by both these elements were suggested. Their coordinates, anisotropic displacement parameters (ADP) and occupancies were constrained. After the refinement of the metal occupancies in *M1* and *M2* positions and refinement of isotropic displacement parameters, we tried to refine ADPs of the heavy atoms (Y, Hf and K). The isotropic displacement parameters of the four O atoms were constrained to be equal before the final cycles of the refinement. The experimental, calculated and difference pattern of the Rietveld refinement is shown in Fig. 2.

Figures

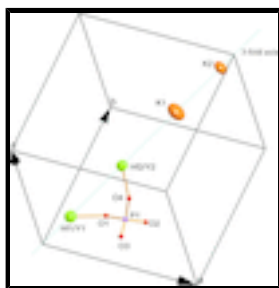


Fig. 1. A view of the asymmetric unit of $\text{K}_2\text{YHf}(\text{PO}_4)_3$. Displacement ellipsoids are drawn at the 50% probability level.

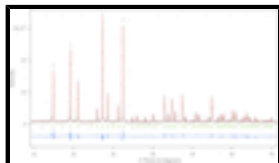


Fig. 2. Rietveld refinement of $\text{K}_2\text{YHf}(\text{PO}_4)_3$. Experimental (dots), calculated (red curve) and difference (blue curve) data for 2θ range 8-72°.

Potassium yttrium hafnium tris(orthophosphate)

Crystal data

$\text{K}_2\text{HfY}(\text{PO}_4)_3$

$M_r = 630.51$

Cubic, $P2_13$

$a = 10.30748$ (9) Å

$b = 10.30748$ (9) Å

$c = 10.30748$ (9) Å

$\alpha = 90^\circ$

$V = 1095.11$ (2) Å³

$Z = 4$

$D_x = 3.824$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.540530$ Å

$T = 293$ K

Specimen form: flat_sheet; particle morphology isometric; colourless

$15 \times 15 \times 1$ mm

$\beta = 90^\circ$

$\gamma = 90^\circ$

Specimen preparation: pressure 101.3 kPa; temperature 293 K

Data collection

Shimadzu XRD-6000 diffractometer

Monochromator: graphite

Specimen mounting: glass container

Specimen mounted in reflection mode

Scan method: step

$T = 293$ K

$2\theta_{\min} = 5.00$, $2\theta_{\max} = 105.00^\circ$

Increment in $2\theta = 0.02^\circ$

Refinement

$R_p = 5.375$

$R_{wp} = 7.075$

$R_{exp} = 2.809$

$R_B = 4.248$

$S = 2.51$

Excluded region(s): undef

Profile function: Thompson–Cox–Hastings pseudo-Voigt Axial divergence asymmetry (Thompson *et al.*, 1987)

48 parameters

14 constraints

Standard least squares refinement

$(\Delta/\sigma)_{\max} = 0.001$

Extinction correction: none

Preferred orientation correction:

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	U_{iso}^*/U_{eq}	Occ. (<1)
K1	0.6984 (4)	0.6984 (4)	0.6984 (4)	0.12 (9)	
K2	0.9307 (5)	0.9307 (5)	0.9307 (5)	0.07 (9)	
Y1	0.14694 (11)	0.14694 (11)	0.14694 (11)	0.06 (9)	0.395 (10)
Y2	0.41559 (18)	0.41559 (18)	0.41559 (18)	0.05 (9)	0.605 (10)
Hf1	0.14694 (11)	0.14694 (11)	0.14694 (11)	0.06 (9)	0.605 (10)
Hf2	0.41559 (18)	0.41559 (18)	0.41559 (18)	0.05 (9)	0.395 (10)
P1	0.4609 (5)	0.2311 (8)	0.1292 (8)	0.06 (9)	
O1	0.3207 (14)	0.2448 (13)	0.0864 (15)	0.07 (9)	
O2	0.5337 (12)	0.3118 (13)	0.0155 (16)	0.07 (9)	
O3	0.4970 (13)	0.0965 (13)	0.1595 (13)	0.07 (9)	
O4	0.4750 (14)	0.3063 (15)	0.2526 (17)	0.07 (9)	

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
K1	0.12 (9)	0.12 (9)	0.12 (9)	−0.031 (4)	−0.031 (4)	−0.031 (4)

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K2	0.07 (9)	0.07 (9)	0.07 (9)	-0.019 (3)	-0.019 (3)	-0.019 (3)
Y1	0.06 (9)	0.06 (9)	0.06 (9)	0.0018 (9)	0.0018 (9)	0.0018 (9)
Y2	0.05 (9)	0.05 (9)	0.05 (9)	0.0034 (9)	0.0034 (9)	0.0034 (9)
Hf1	0.06 (9)	0.06 (9)	0.06 (9)	0.0018 (9)	0.0018 (9)	0.0018 (9)
Hf2	0.05 (9)	0.05 (9)	0.05 (9)	0.0034 (9)	0.0034 (9)	0.0034 (9)

Geometric parameters (Å, °)

K1—O1 ⁱ	2.981 (16)	Hf1—O1 ^{xiii}	2.148 (14)
K1—O2 ⁱⁱ	3.345 (14)	Hf1—O2 ^{xiv}	2.085 (15)
K1—O4 ⁱⁱ	3.413 (16)	Hf2—O3 ⁱ	2.211 (14)
K1—O1 ⁱⁱⁱ	2.981 (16)	Hf2—O4	2.113 (17)
K1—O2 ^{iv}	3.345 (14)	Hf2—O4 ^{xi}	2.113 (17)
K1—O4 ^{iv}	3.413 (16)	Hf2—O3 ⁱⁱⁱ	2.211 (14)
K1—O1 ^v	2.981 (16)	Hf2—O4 ^{xiii}	2.113 (17)
K1—O2 ^{vi}	3.345 (14)	Hf2—O3 ^v	2.211 (14)
K1—O4 ^{vi}	3.413 (16)	Y1—O1	2.148 (14)
K2—O3 ⁱⁱ	2.907 (14)	Y1—O2 ^x	2.085 (15)
K2—O2 ^{vii}	2.912 (14)	Y1—O2 ^{xii}	2.085 (15)
K2—O4 ⁱⁱ	3.207 (17)	Y1—O1 ^{xiii}	2.148 (14)
K2—O4 ^{vii}	3.336 (17)	Y1—O2 ^{xiv}	2.085 (15)
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K2—O2 ^{viii}	2.912 (14)	Y2—O3 ⁱ	2.211 (14)
K2—O4 ^{iv}	3.207 (17)	Y2—O4	2.113 (17)
K2—O4 ^{viii}	3.336 (17)	Y2—O3 ^v	2.211 (14)
K2—O3 ^{vi}	2.907 (14)	Y2—O4 ^{xi}	2.113 (17)
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K2—O4 ^{vi}	3.207 (17)	Y2—O4 ^{xiii}	2.113 (17)
K2—O4 ^{ix}	3.336 (17)	P1—O1	1.518 (16)
Hf1—O1	2.148 (14)	P1—O2	1.621 (17)
Hf1—O2 ^x	2.085 (15)	P1—O3	1.470 (16)
Hf1—O1 ^{xi}	2.148 (14)	P1—O4	1.497 (19)
Hf1—O2 ^{xii}	2.085 (15)		
O1—Hf1—O2 ^x	97.9 (5)	O2 ^x —Y1—O2 ^{xii}	80.1 (5)
O1—Hf1—O1 ^{xi}	89.3 (5)	O1 ^{xiii} —Y1—O2 ^x	172.6 (5)
O1—Hf1—O2 ^{xii}	172.6 (5)	O2 ^x —Y1—O2 ^{xiv}	80.1 (5)
O1—Hf1—O1 ^{xiii}	89.3 (5)	O1 ^{xi} —Y1—O2 ^{xii}	97.9 (5)
O1—Hf1—O2 ^{xiv}	92.5 (5)	O1 ^{xi} —Y1—O1 ^{xiii}	89.3 (5)
O1 ^{xi} —Hf1—O2 ^x	92.5 (5)	O1 ^{xi} —Y1—O2 ^{xiv}	172.6 (5)
O2 ^x —Hf1—O2 ^{xii}	80.1 (5)	O1 ^{xiii} —Y1—O2 ^{xii}	92.5 (5)
O1 ^{xiii} —Hf1—O2 ^x	172.6 (5)	O2 ^{xii} —Y1—O2 ^{xiv}	80.1 (5)
O2 ^x —Hf1—O2 ^{xiv}	80.1 (5)	O1 ^{xiii} —Y1—O2 ^{xiv}	97.9 (5)

O1 ^{xi} —Hf1—O2 ^{xii}	97.9 (5)	O3 ⁱ —Y2—O4	93.1 (6)
O1 ^{xi} —Hf1—O1 ^{xiii}	89.3 (5)	O4—Y2—O4 ^{xi}	87.8 (6)
O1 ^{xi} —Hf1—O2 ^{xiv}	172.6 (5)	O3 ⁱⁱⁱ —Y2—O4	171.5 (6)
O1 ^{xiii} —Hf1—O2 ^{xii}	92.5 (5)	O4—Y2—O4 ^{xiii}	87.8 (6)
O2 ^{xii} —Hf1—O2 ^{xiv}	80.1 (5)	O3 ^v —Y2—O4	83.9 (5)
O1 ^{xiii} —Hf1—O2 ^{xiv}	97.9 (5)	O3 ⁱ —Y2—O4 ^{xi}	83.9 (5)
O3 ⁱ —Hf2—O4	93.1 (6)	O3 ⁱ —Y2—O3 ⁱⁱⁱ	95.4 (5)
O4—Hf2—O4 ^{xi}	87.8 (6)	O3 ⁱ —Y2—O4 ^{xiii}	171.5 (6)
O3 ⁱⁱⁱ —Hf2—O4	171.5 (6)	O3 ⁱ —Y2—O3 ^v	95.4 (5)
O4—Hf2—O4 ^{xiii}	87.8 (6)	O3 ⁱⁱⁱ —Y2—O4 ^{xi}	93.1 (6)
O3 ^v —Hf2—O4	83.9 (5)	O4 ^{xi} —Y2—O4 ^{xiii}	87.8 (6)
O3 ⁱ —Hf2—O4 ^{xi}	83.9 (5)	O3 ^v —Y2—O4 ^{xi}	171.5 (6)
O3 ⁱ —Hf2—O3 ⁱⁱⁱ	95.4 (5)	O3 ⁱⁱⁱ —Y2—O4 ^{xiii}	83.9 (5)
O3 ⁱ —Hf2—O4 ^{xiii}	171.5 (6)	O3 ⁱⁱⁱ —Y2—O3 ^v	95.4 (5)
O3 ⁱ —Hf2—O3 ^v	95.4 (5)	O3 ^v —Y2—O4 ^{xiii}	93.1 (6)
O3 ⁱⁱⁱ —Hf2—O4 ^{xi}	93.1 (6)	O1—P1—O2	100.5 (9)
O4 ^{xi} —Hf2—O4 ^{xiii}	87.8 (6)	O1—P1—O3	113.0 (9)
O3 ^v —Hf2—O4 ^{xi}	171.5 (6)	O1—P1—O4	106.9 (9)
O3 ⁱⁱⁱ —Hf2—O4 ^{xiii}	83.9 (5)	O2—P1—O3	121.4 (8)
O3 ⁱⁱⁱ —Hf2—O3 ^v	95.4 (5)	O2—P1—O4	107.7 (9)
O3 ^v —Hf2—O4 ^{xiii}	93.1 (6)	O3—P1—O4	106.5 (9)
O1—Y1—O2 ^x	97.9 (5)	Hf1—O1—P1	131.7 (9)
O1—Y1—O1 ^{xi}	89.3 (5)	Y1—O1—P1	131.7 (9)
O1—Y1—O2 ^{xii}	172.6 (5)	Hf1 ^{xv} —O2—P1	160.8 (9)
O1—Y1—O1 ^{xiii}	89.3 (5)	Hf2 ^{xvi} —O3—P1	145.6 (9)
O1—Y1—O2 ^{xiv}	92.5 (5)	Hf2—O4—P1	157.5 (10)
O1 ^{xi} —Y1—O2 ^x	92.5 (5)	Y2—O4—P1	157.5 (10)

Symmetry codes: (i) $-x+1, y+1/2, -z+1/2$; (ii) $-x+3/2, -y+1, z+1/2$; (iii) $-z+1/2, -x+1, y+1/2$; (iv) $-y+1, z+1/2, -x+3/2$; (v) $y+1/2, -z+1/2, -x+1$; (vi) $z+1/2, -x+3/2, -y+1$; (vii) $-z+1, x+1/2, -y+3/2$; (viii) $-y+3/2, -z+1, x+1/2$; (ix) $x+1/2, -y+3/2, -z+1$; (x) $x-1/2, -y+1/2, -z$; (xi) z, x, y ; (xii) $-z, x-1/2, -y+1/2$; (xiii) y, z, x ; (xiv) $-y+1/2, -z, x-1/2$; (xv) $x+1/2, -y+1/2, -z$; (xvi) $-x+1, y-1/2, -z+1/2$.

Fig. 1

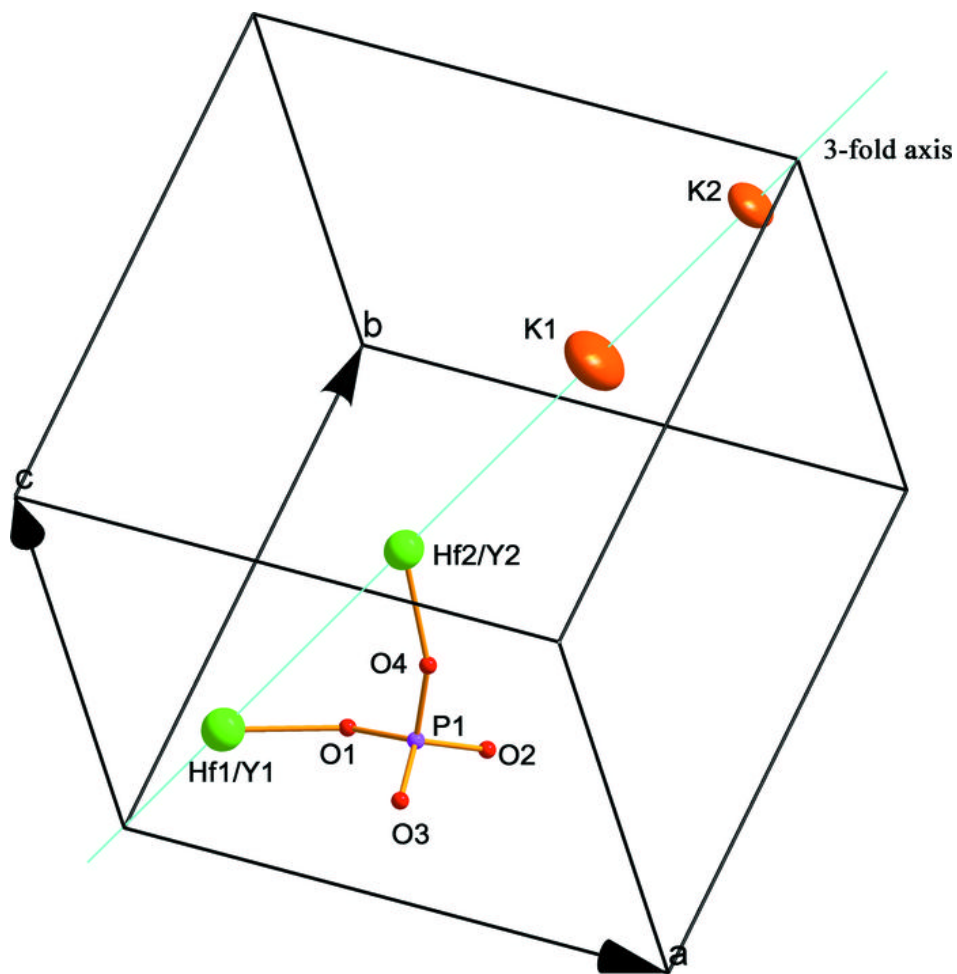


Fig. 2

